

1. IDENTIFICATION OF THE SUBSTANCE / PREPARATION AND COMPANY / UNDERTAKING

Product Name KLEA™ 134a

Manufacturer INEOS Fluor Limited

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Use Subject to Member State regulations, applicable uses are: refrigerant,

blowing agent , propellant , solvent

2. HAZARDS IDENTIFICATION

Low acute toxicity. High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects and asphyxiation. Liquid splashes or spray may cause freeze burns to skin and eyes.

EC Classification Not classified as dangerous according to EC Directive 67/548/EEC.

3. COMPOSITION / INFORMATION ON INGREDIENTS

Alternative names; 1,1,1,2-tetrafluoroethane (HFC 134a)

R 134a

HAZARDOUS INGREDIENT(S)

Hazardous ingredient(s)	%(w/w)	CAS No.	EC No.	EC Classification
1,1,1,2-tetrafluoroethane (HFC 134a)	100	000811-97-2	212-377-0	

4. FIRST-AID MEASURES

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The first aid advice given for skin contact, eye contact, and ingestion is applicable following exposures to the liquid or spray. See also section 11.

Inhalation Remove patient from exposure, keep warm and at rest. Administer

oxygen if necessary. Apply artificial respiration if breathing has ceased or shows signs of failing. In the event of cardiac arrest apply external

cardiac massage. Obtain immediate medical attention.

Skin Contact Thaw affected areas with water. Remove contaminated clothing. Caution:

clothing may adhere to the skin in the case of freeze burns. After contact with skin, wash immediately with plenty of warm water. If irritation or

blistering occur obtain medical attention.

Eye Contact Immediately irrigate with eyewash solution or clean water, holding the

eyelids apart, for at least 10 minutes. Obtain immediate medical attention.

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Ingestion Unlikely route of exposure. Do not induce vomiting. Provided the patient

is conscious, wash out mouth with water and give 200-300 ml (half a pint)

of water to drink. Obtain immediate medical attention.

Further Medical Treatment Symptomatic treatment and supportive therapy as indicated. Adrenaline and similar sympathomimetic drugs should be avoided following exposure

as cardiac arrhythmia may result with possible subsequent cardiac arrest.

FIRE-FIGHTING MEASURES

General HFC 134a is not flammable in air under ambient conditions of

temperature and pressure. Certain mixtures of HFC 134a and air when under pressure may be flammable. Mixtures of HFC 134a and air under

pressure should be avoided.

Certain mixtures of HFCs and chlorine may be flammable or reactive under certain conditions. Thermal decomposition will evolve very toxic

and corrosive vapours. (hydrogen fluoride)

Containers may burst if overheated.

As appropriate for surrounding fire. **Extinguishing Media**

Keep fire exposed containers cool by spraying with water.

Fire Fighting Protective Equipment A self contained breathing apparatus and full protective clothing must be

worn in fire conditions. See Also Section 8.

ACCIDENTAL RELEASE MEASURES

Personal Protection Ensure suitable personal protection (including respiratory protection)

during removal of spillages. See Also Section 8

General Provided it is safe to do so, isolate the source of the leak. Allow small

spillages to evaporate provided there is adequate ventilation.

Large spillages: Ventilate area. Contain spillages with sand, earth or any suitable adsorbent material. Prevent liquid from entering drains, sewers, basements and workpits since the vapour may create a suffocating

atmosphere.

HANDLING AND STORAGE

Handling

Avoid inhalation of high concentrations of vapours. Atmospheric levels should be controlled in compliance with the occupational exposure limit. Atmospheric concentrations well below the occupational exposure limit can be achieved by good occupational hygiene practice.

The vapour is heavier than air, high concentrations may be produced at low levels where general ventilation is poor, in such cases provide adequate ventilation or wear suitable respiratory protective equipment with positive air supply.

Avoid contact with naked flames and hot surfaces as corrosive and very toxic decomposition products can be formed.

Avoid contact between the liquid and skin and eyes.

Avoid venting to atmosphere.

The fluorinated greenhouse gas R 134a may be supplied in returnable containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in containers may not be vented to the atmosphere. Regulation (EC) No. 842/2006 of the European Parliament and the Council on certain fluorinated greenhouse gases.

Process Hazards Liquid refrigerant transfers between refrigerant containers and to and from systems can result in static generation. Ensure adequate earthing. Certain mixtures of HFCs and chlorine may be flammable or reactive

under certain conditions.

Care must be taken to mitigate the risk of developing high pressures in systems caused by a temperature rise when liquid is trapped between closed valves or in cases where containers have been overfilled.

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Storage Keep in a well ventilated place away from fire risk and avoid sources of

heat such as electric or steam radiators.

Avoid storing near to the intake of air conditioning units, boiler units and

open drains.

Specific use Subject to Member State regulations, applicable uses are: refrigerant,

blowing agent, propellant, solvent

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General Wear suitable protective clothing, gloves and eye/face protection. Wear

thermal insulating gloves when handling liquefied gases.

In cases of insufficient ventilation, where exposure to high concentrations of vapour is possible, suitable respiratory protective equipment with

positive air supply should be used.

Eye Protection

Gloves

Occupational Exposure Limits

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Occupational Exposure Limits	CAS No.	LTEL 8	LTEL 8	STEL	STEL	Note:		
		hr	hr	ppm	mg/m³			
		TWA	TWA					
		ppm	mg/m³					
1,1,1,2-Tetrafluoroethane (HFC	000811-97-2	1000	4240	-	-	WEL		
134a)								

9. PHYSICAL AND CHEMICAL PROPERTIES

Form liquefied gas
Colour colourless
Odour slight ethereal
Solubility (Water) slightly soluble

Solubility (Other) Soluble in: alcohols , chlorinated solvents , polyethylene glycol

Boiling Point (° C) -26.2 Melting Point (° C) -101

Vapour Density (Air= 1) 3.66 at normal boiling point

Vapour Pressure (mm Hg) 4270 at 20 ° C Specific Gravity 1.22 at 20 ° C

10. STABILITY AND REACTIVITY

Hazardous Reactions Certain mixtures of HFCs and chlorine may be flammable or reactive

under certain conditions.

Incompatible materials: finely divided metals , magnesium and alloys containing more than 2% magnesium . Can react violently if in contact with alkali metals and alkaline earth metals - sodium , potassium , barium

Hazardous Decomposition Product(s) hydrogen fluoride by thermal decomposition and hydrolysis.

11. TOXICOLOGICAL INFORMATION

Inhalation High exposures may cause an abnormal heart rhythm and prove

suddenly fatal. Very high atmospheric concentrations may cause

anaesthetic effects and asphyxiation.

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Skin Contact Liquid splashes or spray may cause freeze burns. Unlikely to be

hazardous by skin absorption.

Eye Contact Liquid splashes or spray may cause freeze burns.

Ingestion Highly unlikely - but should this occur freeze burns will result.

Long Term Exposure A lifetime inhalation study in rats has shown that exposure to 50000ppm

resulted in benign tumours of the testis. The increased tumour incidence was observed only after prolonged exposure to high levels, and is considered not to be of relevance to humans occupationally exposed to

HFC 134a at or below the occupational exposure limit.

12. ECOLOGICAL INFORMATION

Environmental Fate and Distribution High tonnage material produced in wholly contained systems. High

tonnage material used in open systems. Vapour

Persistence and Degradation

Decomposed comparatively rapidly in the lower atmosphere (troposphere). Atmospheric lifetime is 14 years. Products of

decomposition will be highly dispersed and hence will have a very low concentration. Does not influence photochemical smog (i.e. is not a VOC

under the terms of the UNECE agreement).

Does not deplete ozone.

Has a Global Warming Potential (GWP) of 1300 (relative to a value of 1 for carbon dioxide at 100 years) according to Annex I of Regulation 842/2006 on certain fluorinated greenhouse gases. Values in Annex I are taken from the third assessment report (TAR) of the Intergovernmental

Panel on Climate Change (2001 IPCC GWP values).

United Nations Framework Convention on Climate Change (UNFCCC)

reporting GWP is 1300.

Effect on Effluent Treatment Discharges of the product will enter the atmosphere and will not result in

long term aqueous contamination.

13. DISPOSAL CONSIDERATIONS

Recommended: Best to recover and recycle. If this is not possible, destruction is to be in

an approved facility which is equipped to absorb and neutralise acid

gases and other toxic processing products.

14. TRANSPORT INFORMATION

Hazard Label



Road/Rail

UN No. 3159 ADR/RID Class 2.2

ADR/RID Proper Shipping Name 1,1,1,2-TETRAFLUOROETHANE (REFRIGERANT GAS R 134a)

SEA

IMDG Class 2.3

Marine Pollutant Not classified as a Marine Pollutant

AIR

ICAO/IATA 2.2

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REGULATORY INFORMATION *15.*

European Regulations:

EC Classification Not classified as dangerous according to EC Directive 67/548/EEC.

Special Restrictions: The fluorinated greenhouse gas R 134a may be supplied in returnable

containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in

containers may not be vented to the atmosphere.

Regulation (EC) No. 842/2006 of the European Parliament and the Council on

certain fluorinated greenhouse gases.

Directive 2006/40/EC of the European Parliament and the Council relating to emissions from air-conditioning systems in motor vehicles and amending

Council Directive 70/156/EC.

16. OTHER INFORMATION

This data sheet was prepared in accordance with Directive 2001/58/EC.

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Glossary

WEL: Workplace Exposure Limit (UK HSE EH40)

COM: The company aims to control exposure in its workplace to this limit TLV: The company aims to control exposure in its workplace to the ACGIH limit TLV-C: The company aims to control exposure in its workplace to the ACGIH Ceiling limit The company aims to control exposure in its workplace to the German limit MAK:

Sk: Can be absorbed through skin

Sen: Capable of causing respiratory sensitisation

Biological monitoring guidance value (UK HSE EH40)

The following sections contain revisions or new statements: 1,2,3,7,12,15,16

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